



Solid State Electronics – Unit 1: Atomic Structure

Exercise – Insulator, Conductor and Semiconductor Material KEY

Name: _____

1. The defining property of a semiconductor material is that it can be doped with impurities that alter its electronic properties in a controllable way.
2. The defining property of conductor materials is that it permits electrons to flow freely from particle to particle.
3. The defining property of an insulator is that it does not readily conduct electricity.

Place a 1(semiconductor), 2(conductor) or 3(insulator) in front of each of the following materials listed below to describe the particular material.

2	Silver
	Rubber
1	germanium
3	2 Carbon
	2 Copper
	glass
<hr/>	
2	Gold
	Porcelain
<hr/>	
2	Aluminum
	Plastic
<hr/>	
2	Magnesium
1	selenium
<hr/>	
	Wool
2	Tungsten
2	Water
2	Nickel
	Silk
<hr/>	
2	Mercury
	Paper
2	Platinum
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2	Iron
1	Silicon





**Multi-State
Advanced Manufacturing
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